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1: Which of the following cannot be broken down to anything simpler? A) water: B) table salt: C) silver: D) sugar: 2: Elements combine in fixed mass ratios to form ...

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Let's figure out the **electron configuration** for nickel, right there. 28 **electrons**. We just have to figure out what shells and; orbitals they go in.

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Well what happens is the **electron configuration** of Copper would be 1s22p63s23p64s23d9. However, the **configuration** of having the 3d9 is not what the **electron** â€¦

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Question: The K, L and M shells of an atom are full. Its atomic number is_____.

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22) Which **electron configuration** denotes an atom in its ground state? A) B) C) D) E) 23) The ground state **electron configuration** of Fe is _____. A)1s22s23s23p63d6

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Electron Notations Review - ScienceGeek.net

www.sciencegeek.net/Chemistry/taters/Unit2**Electron**Notations.htm

Which of the following is the correct **electron configuration** notation for the element nitrogen, (N, atomic # 7)? ? 1s 2 2s 2 3s 2 4s 1 ? 1s 2 2s 2 2p 3

1 2 3

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